

New Diterpenes from Alpinia galanga WILD.

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Two new skeleton diterpenes (galanal A and B) and a new labdane type diterpene were isolated from the seeds of Alpinia galanga WILD. (Zingiberaceae) with (E)-8, 17-epoxy-labd-12-ene-15, 16-dial and their structures were established by the spectral evidences.

The rhizomes of A. galanga has been used for flavouring foods in the preparation of meat dishes and curries.¹⁾ In our previous paper,²⁾ the anti-tumour substance and pungent principle were isolated from the rhizomes and the seeds of A. galanga. Continuing with our work on phytochemical constituents of A. galanga, two new skeleton diterpenes (1 and 2)³⁾ and a new labdane type diterpen (3) have been isolated along with (E)-8, 17-epoxy-labd-12-ene-15, 16-dial (4).⁴⁾ Only a few instances have been reported on the isolation of diterpenes from Zingiberaceous plant.^{4,5)} In the present paper, the authors wish to report on the structural elucidation of new diterpenes.

The seeds of A. galanga commercially purchased were extracted with methanol and the methanolic extracts were shaken with n-hexane after addition of water. Chromatographic purification of the n-hexane soluble fraction furnished compound 1-4.

Compound 1, named as galanal A, was obtained as colorless needles, mp 167.0-169.0 °C, molecular formula, C₂₀H₃₀O₃ based on high resolution MS (318.2214). ¹H-NMR (CDCl₃), IR (CHCl₃), and UV (EtOH) spectra indicated the presence of dialdehyde, that is, one was a normal aldehyde (δ 10.10; 1710 cm⁻¹), the other was an unsaturated aldehyde (δ 9.38; 1685 cm⁻¹; 232 nm, ε 7600). Then the presence of a secondary hydroxyl group was indicated by the IR (3620 cm⁻¹)

and $^1\text{H-NMR}(\delta 4.06)$. The remaining $^1\text{H-NMR}$ signals of three quarternary methyl ($\delta 0.77, 0.80, \text{ and } 0.88$) and the $^{13}\text{C-NMR}$ (Table 1) signals were readily accommodated to a labdane skeleton and were similar to published data.⁶⁾ The relationship between two aldehyde groups and a secondary hydroxyl group of **1** was established as shown in partial structure $(-\underset{|}{\text{C}}\text{H}-\text{CH}_2-\text{CH}=\overset{\text{CHO}}{\underset{|}{\text{C}}}-\text{CH}_2-\text{CHOH}-\overset{\text{CHO}}{\underset{|}{\text{C}}}-)$ by detailed double resonance experiments and correlation spectrum in NMR, compared with those of **5** which was afforded by oxidation of **1** with PCC. $^1\text{H-NMR}$ data were shown in Table 2. To combine this partial structure in a labdane skeleton, ring C was considered to be present as a seven-membered ring. The presence of a seven-membered ring can be also explained satisfactorily by the NMR(Table 2) and IR spectra(1700 cm^{-1}) of **5**.

The relative configuration of B/C ring junction and a hydroxyl group was determined by NOE experiments of **1** and **5**. Irradiation of H-20 and H-9 in **5** caused 7 and 12% increase in the integrated intensity of H-17 and H-14, respectively. Therefore the ring junction was concluded to be trans. Furthermore 4% NOE was observed between the protons at C-15 and C-17 in **1**, indicating the relative configuration of a hydroxyl group was to be α .

Compound **2**, named as galanal B, was obtained as colorless needles, mp $134.0-134.5\text{ }^\circ\text{C}$, molecular formula, $\text{C}_{20}\text{H}_{30}\text{O}_3$ (318.2213) and was considered to be closely related structure to **1** by spectral data. Compound **1** and **2** were proved to be an epimer with respect to a hydroxyl group by oxidation of **2** with PCC which afforded the same compound to **5** in all respects. 5% NOE was observed between the protons at C-9 and C-15 in **2**, indicating the relative configuration of a hydroxyl group was to be β .

Compound **3** was obtained as an unstable colorless oil and a major di-terpene, molecular formula, $\text{C}_{20}\text{H}_{30}\text{O}_2$ (302.2238), $[\alpha]_{\text{D}} +15.8^\circ$. The spectral data was completely identical with labda-8(17), 12-diene-15, 16-dial($6: [\alpha]_{\text{D}} -15.0^\circ$) except for the optical rotation, which was isolated from Alpinia speciosa in our previous communication.⁷⁾ To determine the absolute structure, Octant rule was applied to the compound **7**(Chart 2) which was afforded by ozonolysis of **3**. The CD spectrum of **7** showed the negative Cotton effect(289 nm , $\Delta\epsilon = -2.79$, $c = 0.18$, EtOH). Therefore the absolute structure of **3** was determined as shown in Chart 1 and that of **6** isolated from A. speciosa was also concluded as shown in Chart 1.⁸⁾

Table 1. ^{13}C -NMR Chemical Shifts of 1-5 (100MHz in CDCl_3)

No.	1	2	3(6)	4	5
1	38.6	38.9	39.2 ^{a)}	39.6	38.9
2	18.5 ^{a)}	18.7	19.3	18.4	18.6 ^{a)}
3	41.7	41.7	42.0	42.0	41.6
4	33.4	33.4	33.6	33.6	33.4
5	53.6 ^{b)}	55.5 ^{a)}	55.4 ^{b)}	52.8	51.7
6	18.6 ^{a)}	19.1	24.1	20.0	18.8 ^{a)}
7	28.5	34.5	39.3 ^{a)}	39.6	35.4
8	55.5	55.4	148.0	57.6	66.3
9	55.8 ^{b)}	55.6 ^{a)}	56.5 ^{b)}	55.2	55.5
10	38.5	39.0	39.6	40.0	37.7
11	23.6	24.1	24.7	22.4	26.6
12	156.1	157.7	159.8	161.2	153.4
13	142.3	140.8	134.7	136.0	135.4
14	27.7	28.7	37.8	36.0	31.6
15	71.3	78.7	197.1	198.0	206.2
16	193.3	193.5	193.4	194.0	191.7
17	206.5	208.1	107.8	48.8	200.0
18	33.2	33.3	33.6	33.6	33.3
19	21.3	21.4	21.7	21.6	21.3
20	16.6	15.9	14.4	14.8	15.5

a) and b) Tentative.

Table 2. ^1H -NMR Chemical Shifts of 1, 2, and 5 (400MHz in CDCl_3)

H	1	2	3
9	1.61(d, 10.3Hz)	1.47(d, 10.3Hz)	2.34(dd, 2.2, 12.0Hz)
11 α	2.80(m)	3.12(ddddd, 0.9, 1.8, 4.5, 10.3, 17.2Hz)	2.86(dddd, 2.1, 2.2, 6.3, 20.6Hz)
β	2.49(dd, 8.3, 18.6Hz)	2.55(dd, 8.5, 17.2Hz)	2.99(dddd, 3.0, 3.5, 12.0, 20.6Hz)
12	6.89(dd, 2.2, 8.3Hz)	7.04(dd, 4.5, 8.5Hz)	6.87(ddd, 2.5, 3.0, 6.3Hz)
14 α	2.60-2.70(m)	2.92(ddd, 0.9, 9.0, 16.4Hz)	3.68(dddd, 2.1, 2.5, 3.5, 14.1Hz)
β		2.67(ddd, 1.8, 1.8, 16.4Hz)	3.53(d, 14.1Hz)
15	4.06(dd, 2.9, 8.9Hz)	3.54(dd, 1.8, 9.0Hz)	
16	9.38(s)	9.40(s)	9.40(s)
17	10.10(s)	10.20(s)	9.86(s)
18	0.80(s) ^{a)}	0.77(s) ^{a)}	0.81(s) ^{a)}
19	0.88(s) ^{a)}	0.86(s) ^{a)}	0.90(s) ^{a)}
20	0.77(s)	0.78(s)	0.77(s)

a) The assignments may be reversed.

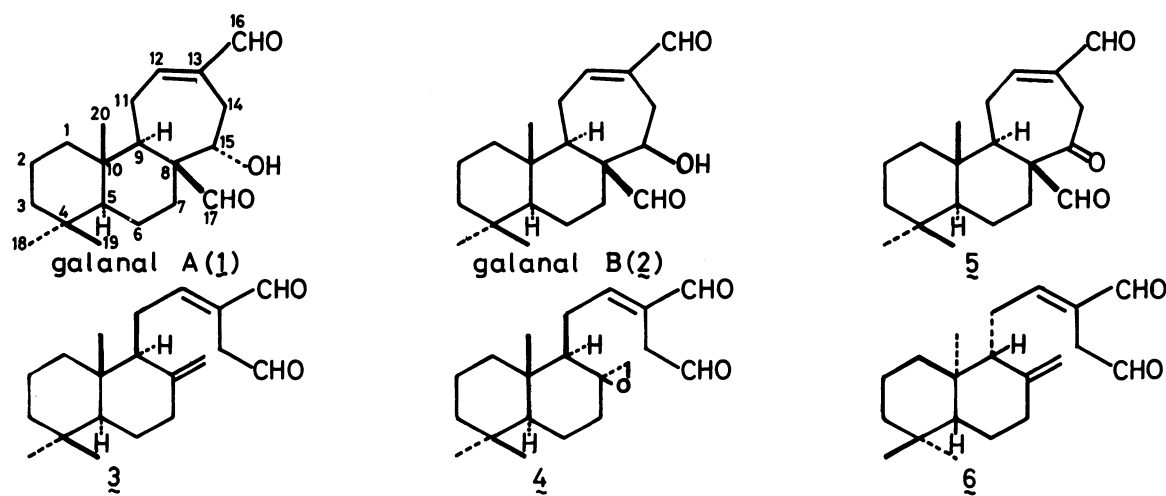


Chart 1.

Compound 4 was completely identical with (E)-8, 17-epoxy-labd-12-ene-15, 16-dial⁴⁾ which was an antifungal labdane dialdehyde.⁹⁾

In view of biogenesis, galanal A and B were considered to be derived by ring closure following the cleavage of epoxide of 4.

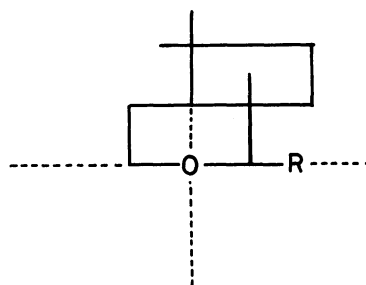
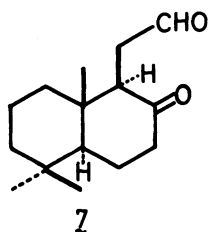


Chart 2.

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- 3) galanal A: $[\alpha]_D -44.0^\circ$ (c 0.1, CHCl_3). IR(CHCl_3): 3620, 2940, 1710, 1685, 1640, 1460, 1445, 1390, 1370, 1165, 1055.
galanal B: $[\alpha]_D -48.0^\circ$ (c 0.1, CHCl_3). IR(CHCl_3): 3610, 2950, 1710, 1680, 1650, 1460, 1440, 1390, 1370, 1160, 1085, 1050. UV(EtOH): 236 nm, ϵ 8000.
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